

FORM PTO-1199 (Modified)  
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U.S. DEPARTMENT OF COMMERCE PATENT AND TRADEMARK OFFICE

ATTORNEY'S DOCKET NUMBER

TRANSMITTAL LETTER TO THE UNITED STATES  
DESIGNATED/ELECTED OFFICE (DO/EO/US)  
CONCERNING A FILING UNDER 35 U.S.C. 371

4004-028-30

U.S. APPLICATION NO. (IF KNOWN, SEE 37 CFR

10/031237

INTERNATIONAL APPLICATION NO.

PCT/EP00/05615

INTERNATIONAL FILING DATE

19 JUNE 2000

PRIORITY DATE CLAIMED

20 JULY 1999

TITLE OF INVENTION

PYROLYTIC LAYER OF ALUMINIUM OXYNITRIDE AND GLAZING COMPRISING SAME

APPLICANT(S) FOR DO/EO/US

LEGRAND, Philippe

Applicant herewith submits to the United States Designated/Elected Office (DO/EO/US) the following items and other information:

1. ☒ This is a **FIRST** submission of items concerning a filing under 35 U.S.C. 371.
2. ☐ This is a **SECOND** or **SUBSEQUENT** submission of items concerning a filing under 35 U.S.C. 371.
3. ☒ This is an express request to begin national examination procedures (35 U.S.C. 371(f)). The submission must include items (5), (6), (9) and (24) indicated below.
4. ☒ The US has been elected by the expiration of 19 months from the priority date (Article 31).
5. ☒ A copy of the International Application as filed (35 U.S.C. 371 (c) (2))
  - a. ☐ is attached hereto (required only if not communicated by the International Bureau).
  - b. ☒ has been communicated by the International Bureau.
  - c. ☐ is not required, as the application was filed in the United States Receiving Office (RO/US).
6. ☒ An English language translation of the International Application as filed (35 U.S.C. 371(c)(2)).
  - a. ☒ is attached hereto.
  - b. ☐ has been previously submitted under 35 U.S.C. 154(d)(4).
7. ☒ Amendments to the claims of the International Application under PCT Article 19 (35 U.S.C. 371 (c)(3))
  - a. ☐ are attached hereto (required only if not communicated by the International Bureau).
  - b. ☐ have been communicated by the International Bureau.
  - c. ☐ have not been made; however, the time limit for making such amendments has NOT expired.
  - d. ☒ have not been made and will not be made.
8. ☐ An English language translation of the amendments to the claims under PCT Article 19 (35 U.S.C. 371(c)(3)).
9. ☐ An oath or declaration of the inventor(s) (35 U.S.C. 371 (c)(4)).
10. ☐ An English language translation of the annexes to the International Preliminary Examination Report under PCT Article 36 (35 U.S.C. 371 (c)(5)).
11. ☐ A copy of the International Preliminary Examination Report (PCT/IPEA/409).
12. ☒ A copy of the International Search Report (PCT/ISA/210).

## Items 13 to 20 below concern document(s) or information included:

13. ☐ An Information Disclosure Statement under 37 CFR 1.97 and 1.98.
14. ☐ An assignment document for recording. A separate cover sheet in compliance with 37 CFR 3.28 and 3.31 is included.
15. ☒ A **FIRST** preliminary amendment.
16. ☐ A **SECOND** or **SUBSEQUENT** preliminary amendment.
17. ☐ A substitute specification.
18. ☐ A change of power of attorney and/or address letter.
19. ☐ A computer-readable form of the sequence listing in accordance with PCT Rule 13ter.2 and 35 U.S.C. 1.821 - 1.825.
20. ☐ A second copy of the published international application under 35 U.S.C. 154(d)(4).
21. ☐ A second copy of the English language translation of the international application under 35 U.S.C. 154(d)(4).
22. ☐ Certificate of Mailing by Express Mail
23. ☒ Other items or information:

## Verification of Translation

Copy of the published International Application (WO 01/05723), including English language abstract and International Search Report

White Advance Serial Number Card

24. The following fees are submitted..

**BASIC NATIONAL FEE (37 CFR 1.492 (a) (1) - (5)) :**

- ☐ Neither international preliminary examination fee (37 CFR 1.482) nor international search fee (37 CFR 1.445(a)(2)) paid to USPTO and International Search Report not prepared by the EPO or JPO ..... \$1040.00
- ☒ International preliminary examination fee (37 CFR 1.482) not paid to USPTO but International Search Report prepared by the EPO or JPO ..... \$890.00
- ☐ International preliminary examination fee (37 CFR 1.482) not paid to USPTO but international search fee (37 CFR 1.445(a)(2)) paid to USPTO ..... \$740.00
- ☐ International preliminary examination fee (37 CFR 1.482) paid to USPTO but all claims did not satisfy provisions of PCT Article 33(1)-(4) ..... \$710.00
- ☐ International preliminary examination fee (37 CFR 1.482) paid to USPTO and all claims satisfied provisions of PCT Article 33(1)-(4) ..... \$100.00

**ENTER APPROPRIATE BASIC FEE AMOUNT =****CALCULATIONS PTO USE ONLY**

Surcharge of \$130.00 for furnishing the oath or declaration later than months from the earliest claimed priority date (37 CFR 1.492 (e)).

☐ 20 ☒ 30

\$890.00

\$130.00

**CLAIMS****NUMBER FILED****NUMBER EXTRA****RATE**

Total claims 9 - 20 = 0 x \$18.00 \$0.00

Independent claims 1 - 3 = 0 x \$84.00 0

Multiple Dependent Claims (check if applicable). ☐

\$0.00

**TOTAL OF ABOVE CALCULATIONS =**

\$1,020.00

☒ Applicant claims small entity status. See 37 CFR 1.27). The fees indicated above are reduced by 1/2.

\$0.00

**SUBTOTAL =**

\$1,020.00

Processing fee of \$130.00 for furnishing the English translation later than months from the earliest claimed priority date (37 CFR 1.492 (f)).

☐ 20 ☐ 30 +

\$0.00

**TOTAL NATIONAL FEE =**

\$1,020.00

Fees for recording the enclosed assignment (37 CFR 1.21(h)). The assignment must be accompanied by an appropriate cover sheet (37 CFR 3.28, 3.31) (check if applicable). ☐

\$0.00

**TOTAL FEES ENCLOSED =**

\$1,020.00

Amount to be:  
refunded \$  
charged \$

- a. ☒ A check in the amount of \$1,020.00 to cover the above fees is enclosed.
- b. ☐ Please charge my Deposit Account No. \_\_\_\_\_ in the amount of \_\_\_\_\_ to cover the above fees. A duplicate copy of this sheet is enclosed.
- c. ☒ The Commissioner is hereby authorized to charge any additional fees which may be required, or credit any overpayment to Deposit Account No. 50-1442. A duplicate copy of this sheet is enclosed.
- d. ☐ Fees are to be charged to a credit card. **WARNING:** Information on this form may become public. Credit card information should not be included on this form. Provide credit card information and authorization on PTO-2038.

**NOTE:** Where an appropriate time limit under 37 CFR 1.494 or 1.495 has not been met, a petition to revive (37 CFR 1.137(a) or (b)) must be filed and granted to restore the application to pending status.

SEND ALL CORRESPONDENCE TO:

Supervisor, Patent Prosecution Services  
PIPER MARBURY RUDNICK & WOLFE, LLP  
1200 Nineteenth Street, NW  
Washington, DC 20036-2412  
US

SIGNATURE

SCHNEIDER, Jerold I.

NAME

24,765

REGISTRATION NUMBER

17 JANUARY 2002

DATE

Docket No. 4004-028-30

PATENT COOPERATION TREATY (PCT)  
IN THE UNITED STATES DESIGNATED/ELECTED OFFICE (DO/EO/US)

In re application of: LEGRAND, Philippe  
Int'l Application No.: PCT/EP00/05615  
Int'l Filing Date: 19 June 2000  
Priority Date: 20 July 1999  
U.S. Application No.: New U.S. National Stage Application  
For: PYROLYTIC LAYER OF ALUMINIUM OXYNITRIDE AND GLAZING  
COMPRISING SAME

PRELIMINARY AMENDMENT

Assistant Commissioner for Patents  
Box PCT  
Washington, D.C. 20231

Sir:

With reference to the accompanying English translation  
and prior to any action on the merits, please amend this  
application as follows:

IN THE SPECIFICATION

Page 1, between lines 2 and 3, insert the following including  
a section heading:

-- CROSS REFERENCE TO RELATED APPLICATIONS

This application is the U.S. national stage of  
International Application No. PCT/EP00/05615 filed 19 June  
2000, and Luxembourg Application No. 90420 filed 20 July 1999.  
The entirety of each of those applications is incorporated  
herein by reference.

BACKGROUND OF THE INVENTION --

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Page 4, between lines 2 and 3 insert the section heading:

-- SUMMARY OF THE INVENTION --

Page 12, before line 1, insert the section heading:

-- DETAILED DESCRIPTION --

**IN THE CLAIMS**

Please cancel Claims 1-13 without prejudice or disclaimer and add the following new Claims 14-22.

-- 14. (New) Glazing comprising a glass substrate coated with a layer of aluminium oxynitride deposited by gas-phase pyrolysis, the thickness and refractive index characteristics thereof being selected so as to attenuate the reflected colours produced by an oxide layer providing the glazing with low-emission and/or solar protection properties, said layer being deposited onto the aluminium oxynitride layer.

15. (New) Glazing according to Claim 1, and including at least one of the following features (A) through (D);

(A) wherein the constituent elements of the aluminium oxynitride layer are respectively in the following atomic proportions:

Al	from 40 to 50%
N	from 20 to 50%
O	from 10 to 60%;

(B) wherein the refractive index of the aluminium oxynitride layer is in the range of between 1.6 and 1.8;

(C) wherein the thickness of the aluminium oxynitride layer is in the range of between 500 and 900 ångströms; and

(D) wherein the oxide layer providing the low-emission and/or solar protection properties is a layer based on doped tin oxide.

16. (New) Glazing according to Claim 15 and including at least two of the features (A) through (D).
17. (New) Glazing according to Claim 15 and including all of the features (A) through (D).

18. (New) Glazing according to Claim 17 wherein:  
the constituent elements of the aluminium oxynitride layer are respectively in the following atomic proportions:

Al	from 45 to 50%
N	from 22 to 30%
O	from 20 to 27%;

the refractive index of the aluminium oxynitride layer is in the range of between 1.65 and 1.75; and

the aluminium oxynitride layer has a thickness in the range of between 650 and 850 ångströms.

19. (New) Glazing according to Claim 14 wherein the oxide layer providing the low-emission and/or solar protection properties is a layer based on at least one of the following (E) through (G):
- (E) doped tin oxide;
  - (F) is a tin oxide layer containing antimony oxide, the atomic ratio Sb/Sn being in the range of between 0.02 and 0.15;
  - (G) fluorine-doped tin oxide.
20. (New) Process for the production of glazing according to Claim 14, wherein the aluminium oxynitride layer is formed by pyrolysis using gaseous precursors comprising aluminium trichloride or trimethyl aluminium.

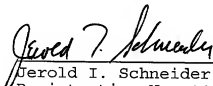
21. (New) Process for the production of glazing according to Claim 20, wherein the gaseous precursors also comprise ammonia.
22. (New) Process for the production of glazing according to Claim 20, wherein, when the aluminium precursor is aluminium trichloride, the precursors also contain water vapour. --

REMARKS

This is the entry into the U.S. National Stage of International Application No. PCT/EP00/05615 filed 19 June 2000. The foregoing amendments are to place this application into U.S. format. Claims 14 through 22 are pending.

Applicant respectfully submits that, in view of the foregoing amendments and remarks, the application is in condition for examination. Favorable consideration is respectfully requested.

Respectfully submitted,  
PIPER MARBURY RUDNICK & WOLFE LLP

  
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Jerold I. Schneider  
Registration No. 24,765  
Attorney for Applicant

1200 Nineteenth Street, N.W.  
Washington, D.C. 20036-2412  
Telephone (202) 861-3900  
Facsimile (202) 223-2085

3602756


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531 Rec'd PCT/PT 17 JAN 2002

## VERIFICATION OF ENGLISH LANGUAGE TRANSLATION

I, Guy Farmer, state that I am a European Patent Attorney,  
that I am conversant with the French and English Languages,  
and that annexed is a true and accurate English language  
translation of the French Language application PCT/EP00/05615.

Executed at Jumet, Belgium, under penalty of perjury of the  
laws of the United States on this 15 day of January, 2002.

  
Guy Farmer

10031237-020102

Docket No.: 4004-028-30

PYROLYTIC LAYER OF ALUMINIUM OXYNITRIDE AND GLAZING  
COMPRISING SAME

5 The invention relates to thin layers deposited on a transparent glass substrate to modify their optical characteristics. In particular, the invention concerns thin layers interposed between the glass substrate and another thin functional layer, in particular a solar protection or low-emission layer.

10 It is well known that the presence of a low-emission or solar protection layer, in particular those with the most suitable thicknesses and refractive indexes, both with respect to efficiency and to industrial application, causes parasitic colorations, in particular in reflected light. These colorations which occur in the form of areas  
15 of iridescence must be fully suppressed for conventional uses, and most particularly for clear glasses where their presence is most detrimental to the appearance of those glazing units comprising such glasses.

In order to prevent these phenomena, it is well  
20 known to interpose a "sub-layer", the refractive index and thickness of which are selected to attenuate or practically eliminate these areas of iridescence, between the functional layer and the glass substrate. Theory allows the appropriate values for these parameters to be  
25 determined precisely. However, in practice some difficulties are encountered in the implementation because of the many technical requirements which



industrial production must meet to produce these sub-layers in an efficient and economical manner.

While various solutions have been proposed for the production of these sub-layers, the most widely used is that of silicon-based sub-layers. The most advantageous methods of deposition from the point of view of expense for the formation of these sub-layers are based on pyrolysis, and most particularly gas-phase pyrolysis (CVD). In these methods the deposition of silicon working from gaseous precursors such as silanes is generally conducted directly on the glass ribbon continuously while this is being produced. The temperature conditions of the glass enable the reaction of pyrolysis of the precursors which are brought into contact with this. However, the choice of proceeding in these conditions implies that there are set requirements associated with the special features of the assembly into which the deposition operation has just been integrated.

The deposition of sub-layers in the conditions which have just been outlined must be conducted quickly. The ribbon of glass to be coated passes under ducts which bring the precursors into contact with it. The contact time of the gas and glass, irrespective of the arrangements of the plant, is necessarily limited. Therefore, highly reactive precursors must be used to achieve the thickness of deposition during this very short period. The precursors of silicon layers have a very high reactivity in temperature conditions such as

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those of the ribbon of glass exiting from the molten tin bath.

However, previous methods of forming silicon-based sub-layers have some disadvantages. In particular, the  
5 precursors of silicon cause rapid fouling of the plant.

The deposition ducts are disposed in direct proximity to the ribbon of glass to ensure good contact and thus a favourable output from the reaction. The distance separating the ribbon of glass from the duct or  
10 ducts is generally only a few mm. In these conditions, parasitic deposits of reagents on the structures in the vicinity of the duct rapidly reach dimensions which necessitate cleaning, otherwise the presence of parasitic deposits causes disruptions in the flows of gas, which  
15 lead to irregularities in the layer. If fouling continues for long the glass may even rub against the concretions formed which leads to unacceptable marks and scratches. In the present state, cleaning constitutes an interruption in production at regular intervals, and this  
20 is even more detrimental to the good function of the operation since these cleaning operations are very frequent.

Moreover, the sub-layer must have well controlled properties, in particular the refractive index, in order  
25 to take into account the special features of the layers from which they eliminate the disadvantages described above. The precise characteristics of the silicon-based layers are difficult to control. Various proposals have

been made to achieve a well defined refractive index without having come to a perfectly satisfactory method.

For the reasons indicated above, it is desirable to be able to have new sub-layers which better meet practical needs. It is the aim of the invention to provide such sub-layers. In particular the invention proposes to provide sub-layers, whose production can be well controlled, even when using CVD, without this control being prejudicial to the speed of the reaction.

The invention proposes to dispose as sub-layer on a glass substrate a transparent layer based on aluminium oxynitride deposited by a gas-phase pyrolysis technique. The designation aluminium oxynitride refers to the product based on aluminium, nitrogen and oxygen in variable proportions which are specified precisely below. The analysis of layers using a scanning electron microscope (SEM) does not reveal any fault in the regularity at resolutions of less than a tenth of a micron. For the envisaged applications, the layer is thus presented as a part of a perfectly homogeneous composition, irrespective of its atomic formula of the type  $AlN_xO_y$ , wherein x and y can vary appreciably as indicated further below.

The use of thin layers based on aluminium nitrides has been the subject of numerous publications. The object of the main applications of this type of layers is to improve the wear resistance properties of the substrate on which they are deposited. These layers are relatively

thick and do not have any particular optical characteristics.

Other layers based on aluminium nitrides have been proposed to be applied to glazing units to provide protection to an underlying functional, mainly reflective, layer. For example, it is a matter of protecting a layer based on TiN or Ag from oxidation.

It is also proposed, in the case of a vacuum deposition technique, to use a layer, in particular based on aluminium nitride, on the glass substrate in order to protect a titanium nitride layer subsequently deposited. In this case the layer has a very reduced thickness, a few nanometres at maximum (EP-A 536607).

Previous literature reports on the use of assemblies of reflective layers, particularly metal layers, joined to absorbent layers of various types, of aluminium nitride, the whole being vacuum deposited to form glazing units with solar protection properties.

Aluminium oxide layers are known for their properties of mechanical hardness and their chemical inertia. Studies have also been conducted to form thin layers by gas-phase pyrolysis for which the refractive index has been determined (Journal of Electronic Material Vol. 17, No. 6, pp. 509-517). The refractive indexes of the layers obtained range from 1.62 to 1.63. These indexes are less than those which are preferred to produce the attenuation sought according to the invention at least when a single sub-layer is used. The literature

also proposes forming assemblies comprising at least two sub-layers with graduated indexes to prevent areas of iridescence. Moreover, previous aluminium oxide layers have been produced in very specific conditions in the laboratory. The reactions for forming these layers using trimethyl aluminium (TMA) are very difficult to control. Even in the laboratory, the reaction develops very significantly in the gas phase and the resulting aluminium powder must be subject to specific measures to prevent it from being deposited on the glass plate serving as substrate in the form of a mass foreign to the layer. If one adds that the speed of the formation of the layer remains relatively low, irrespective of the precautions taken (at best some nanometres per second), it is found that such layers far from satisfy the requirements outlined above.

The invention proposes to use layers based on aluminium oxynitride produced by gas-phase pyrolysis on a glass substrate so as to attenuate the reflected colours produced by an oxide layer providing the glazing with low-emission and/or solar protection properties, said layer being deposited onto the aluminium oxynitride layer.

The aluminium oxynitride layers used according to the invention have variable compositions which are dependent on the conditions of application. Advantageously, in particular to obtain the most

satisfactory refractive indexes, the atomic proportions of the constituents of these layers are as follows:

	Al	from 40 to 50%
	N	from 20 to 50%
5	O	from 10 to 60%.

Particularly preferred, the composition of the layers according to the invention is:

	Al	from 45 to 50%
	N	from 22 to 30%
10	O	from 20 to 27%.

The aluminium oxynitride layers according to the invention have a variable refractive index depending on their composition and the conditions under which they were obtained. This is between 1.6 and 1.8. For the  
15 envisaged use, i.e. as sub-layer of a low-emission and/or solar protection layer, the layer must have an intermediate index between that of the substrate and that of the layer superposed on it. Preferably, the refractive index of the aluminium oxynitride layer is in the range  
20 of between 1.65 and 1.75, particularly when the low-emission or solar protection layer is a tin oxide-based layer.

In order to obtain the anti-iridescence effect, the layers according to the invention must have a thickness  
25 which depends partly on its refractive index and that of the substrate and of the low-emission or solar protection layer. As a function of this refractive index, the thickness of the layers is advantageously in the range of

between 50 and 90 nanometres, and preferably between 65 and 85 nanometres. The layers according to the invention can contain elements other than Al, N and O in small proportions. These elements are normally impurities introduced with the precursors or residues of pyrolysis reactions, or may also come from the substrate itself.

In the production of the layers, when aluminium trichloride is used as precursor, a possible residual element is chlorine. The presence of chlorine in the layer is not desirable. The residual content of chlorine in the layers according to the invention is as low as possible and advantageously remains less than 4%.

The layer according to the invention may also contain alkaline elements, in particular sodium coming from the substrate via diffusion. Experience has shown that the sodium contents remain very low and normally are less than 1%. As we shall explain further below, this low content results in resistant layers in the traditional tests conducted on this type of layers.

The layers according to the invention are deposited by gas-phase pyrolysis. Advantageously, this is conducted directly on the ribbon of glass inside the "float" bath, or immediately on exiting from this. In these conditions the operation is conducted continuously.

The choice of operating directly on the production line largely determines the temperature conditions of the pyrolysis. In practice, the temperature is normally in

the range of between 600 and 730°C at the locations indicated.

The precursors used must meet a range of requirements. They must be able to react satisfactorily in the working conditions. Their reactivity must be sufficient to be able to produce relatively thick layers in a very short time. However, this must not be such that it leads to unwanted parasitic reactions such as reactions in the gas phase before contact with the glass.

The precursors must also be capable of being easily put into gaseous form to be transported to the substrate in conditions easily achieved industrially. For liquid or solid compositions at ordinary temperature it is important that vaporisation is sufficient at a temperature which is not too high to permit transport to the pyrolysis ducts without the risk of condensation in the duct systems.

Other requirements also apply to the precursors. It goes without saying that these products must be as inexpensive as possible, even if they are used in very low quantities. They must not have a toxicity which renders them unusable in the industrial domain in the usual envisaged processing conditions.

Taking these requirements into account, the inventors have shown that it is advantageous to use aluminium trichloride. Trimethyl aluminium (TMA) which is a potential precursor favourably meets the reactivity requirements. It has the disadvantage of high



flammability. Therefore, its use requires more restrictive measures to be taken.

Reference is made below to aluminium trichloride, but this does not exclude use according to the invention  
5 of other precursors of aluminium, in particular TMA.

Aluminium trichloride is sufficiently volatile when the temperature exceeds 150°C. It is sensitive to water and must therefore be placed or prepared in an essentially anhydrous medium. In practice, it is  
10 advantageous according to the invention to proceed with production of aluminium trichloride at the time of its use. This is achieved in a chlorinator-type arrangement.

The formation of nitride also includes the use of a nitrogenous reagent. With aluminium chloride the  
15 inventors advantageously use ammonia, nitrogen fluoride  $\text{NF}_3$ , or amines. The amines which are usable for the production of layers according to the invention preferably include amines with a relatively low molecular mass such as methylamine, ethylamine, propylamine and  
20 dimethylamine. The choice of one or other of the nitride precursors allows the reactivity to be adjusted, in particular as a function of the pyrolysis temperature.

The precursors are mixed with a carrier gas which is inert with respect to the precursors in the conditions of  
25 the reaction. Advantageously, this carrier gas is nitrogen.

When aluminium trichloride is used as precursor, the presence of water has an influence on the composition and

optical characteristics of the layer. The water content is one of the factors which determine the proportions of oxygen and nitrogen in the formed layer, and as a result the refractive index of the layer. It is therefore  
5 possible to adjust the nature and characteristics by way of the water content.

The sub-layers of aluminium oxynitride are advantageously used in glazing units in which the low-emission or solar protection layer is an oxide layer of  
10 tin oxide, indium oxide, etc. According to practice in this field, these oxides are also advantageously doped. Tin oxide may thus be doped with fluorine and indium oxide may be doped with tin. Similarly, it may also be a layer of tin containing antimony in accordance with the  
15 instruction of publications BE-A1010321 and 1010322.

For the glazing units comprising a layer of tin containing antimony, the atomic proportion of Sb/Sn is advantageously in the range of between 0.02 and 0.15, and preferably between 0.05 and 0.12%.

20 Advantageously, the functional layer is likewise deposited by a gas-phase pyrolysis method. In this case, it is preferred that the deposition of the sub-layer is followed immediately by that of the functional layer. When the sub-layer is produced inside the float bath, the  
25 deposition of the layer can follow immediately, either in the float or upon exit therefrom. The temperature and ambient conditions permit these two combinations.

The invention is described in detail in the following practical examples.

In all the examples given, the precursor of aluminium is aluminium trichloride. It is vaporised by sublimation at 150°C (145°C for example 2), and entrained in a flow of nitrogen at this same temperature. The chlorine content in the nitrogen is fixed at 1% and the flow rate of nitrogen is 5 standard litres per minute.

Some ammonia is also mixed with nitrogen as carrier gas. The proportion of ammonia used is much higher than that theoretically necessary to react with chlorine. In the following example, with the exception of example 3, the flow rate of ammonia is regulated in the pilot plant to 0.5 of a standard litre (SL) per minute. The flow rate of carrier nitrogen is 20 standard litres per minute, except also for example 3. For the latter, the respective flow rates are 15 SL of nitrogen and 5 SL of ammonia.

The water content is extremely low. In examples 1 to 8, it is equivalent to a flow rate of 0.02 SL per minute and 0.13 SL for example 9.

In all the examples the contact time of the gas with the hot glass is 5 seconds.

The following table shows the following for each test:

- the temperature of the glass;
- the average index  $n_r$  measured by reflectometer;
- the thickness  $e_r$  determined by reflectometer;

- the index  $n_e$  determined by ellipsometer for the wavelength 550 nm;
- the thickness  $e_e$  determined by ellipsometer;
- the atomic percentages of constituents of the layers.

N°	t C °	$n_R$	$e_R$ (nm)	$n_e$ 550nm	$e_e$ (nm)	Al(at%)	N(at%)	O(at%)
1	700	1.69	75	1.71	79.7	47	23	25
2	700	1.71	75	-	-	-	-	-
3	700	1.69	115	1.66	129.4			
4	730	1.76	75	1.76	76.8	48	28	18
5	680	1.68	67	1.69	68.1	46	22	30
6	650	1.67	67	-	-	46	17	35
7	600	-	-	-	-	-	-	-
8	700	1.71	88	1.71	80.5	48	26	22
9	700	-	14	-	-	47	17	32

It must firstly be noted that the layer produced on the glass at 600°C does not allow adequate determination of its characteristics. For the envisaged production, this temperature value constitutes, in the test conditions, the limit below which it is not possible to have an acceptable layer. The layer is too cloudy, taking into account the required transparency.

Looking at the different temperature used, it is found that irrespective of the method used, the index increases with the pyrolysis temperature. In the reflectometer it passes from 1.67 at 650°C to 1.76 at 730°C; or in the ellipsometer from 1.44 to 1.76 for the same temperatures. The reasons for this development are not completely clear. However, it is found that these differences also follow the differences in the proportions of the constituent elements, which would

indicate that the reactive mechanisms are displaced by the increase in temperature.

The analyses, the results of which also appear in the table, show that the nitrogen content of the layer  
5 increases with the temperature.

Conversely, if the water content is increased, as in example 9, this benefits the presence of oxygen in the layer and the index tends to decrease.

The structure of the layer is not made perfectly  
10 clear. The diffraction X shows the presence of a structure with the crystalline appearance of AlN. The microscopic study of the layers, as indicated above, does not reveal any defect in homogeneity at the resolutions used. Irrespective, what is important for implementation  
15 of the invention is to have a means of regulating the index of the sub-layer by way of the temperature.

It is also found that the rate of deposition increases with the temperature passing through a maximum at a temperature in the order of 700°C. Overall, with the  
20 exception of examples 7 and 9, the rate of deposition is at a satisfactory magnitude to be able to produce layers with the index and thickness suitable for attenuation of areas of iridescence.

The influence of the concentration of ammonia is  
25 approached in example 3. In this example, the very significant excess of this precursor in relation to that of aluminium principally leads to an appreciable increase in the rate of deposition. The overall nature of the

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layer does not appear to be modified, and the refractive index is practically unchanged in relation to that of the deposition serving as reference conducted at the same temperature with the proportions of precursors.

5        Glazing units according to the invention comprising a sub-layer of aluminium oxynitride and a solar protection layer have also been produced. In these tests, the functional layer is a layer of tin containing antimony. These layers are of the type of those described  
10 in the above-mentioned Belgian patent publications.

A first series of tests compares the solar protection layers formed directly on the glass and those formed on the sub-layer of aluminium oxynitride. In these tests, the sub-layer is deposited at 700°C in the  
15 conditions of example 8 above. The deposition of the layer of tin doped with antimony is conducted immediately after formation of the sub-layer at a temperature of 600°C by means of precursors  $\text{SnCl}_4$  and  $\text{SbCl}_5$ . The reaction is conducted in the presence of water vapour.

20        To obtain perfectly identical conditions for deposition of the layer of tin containing antimony, half the glass is masked during deposition of the sub-layer and all the glass is exposed during formation of the layer.

25        Surprisingly, it is found that the two parts of these layers of tin containing antimony are not identical. The part of the layer deposited on the sub-layer is thicker than that formed directly on the glass.

In other words, the presence of the sub-layer seems to facilitate the deposition of the tin layer. This improvement is in the order of 10%. For information purposes, a simultaneous deposition on a glass sheet partly coated with a sub-layer of aluminium oxynitride results in respective thicknesses of 250 and 286 nanometres. Other than the variation in thickness, the properties of the two layers of tin containing antimony are identical. In particular, the presence of the sub-layer does not lead to the formation of cloudiness and attenuates areas of iridescence which show on the part which only contains the tin oxide containing antimony.

In a second series of tests, different sub-layers have been produced on which a layer of tin oxide and antimony is applied, again by gas-phase pyrolysis. The tin oxide and antimony layer is systematically 350 nanometres thick. The thicknesses and indexes of the sub-layers are indicated, as well as the colorimetric indexes in reflected light according to the CIE for each glazing formed. The results are given in the following table:

	$n_c$	$e_c$ (nm)	a	b
10	1.71	75	-3.93	-0.93
11	1.70	80	-2.33	2.33
12	1.72	75	1.28	1.64
13	1.71	73	-7.78	3.85

The results show that the attenuation of parasitic reflections by means of the sub-layer of aluminium oxynitride is comfortably assured by choosing the thickness and index of the sub-layer by way of the

deposition temperature, but also by modifying the proportions of the precursor reagents. A value which is mutually regarded as acceptable for attenuation meets the condition  $(a^2+b^2)^{1/2} < 10$ . This condition is met for each  
5 of the examples indicated in the table. However, it is noted that this condition, and therefore the attenuation, are very sensitive to the characteristics of the layer.

Moreover, the favourable corrosion behaviour and abrasion behaviour of the glazing units according to the  
10 invention have been confirmed as well as their favourable behaviour in the cambering and quenching treatments.

Thanks to the invention, we have a satisfactory alternative to previous methods for the production of low-emission/solar protection glazing units which do not  
15 exhibit the undesirable phenomena of iridescence in reflected light. It is possible to precisely control the characteristics of the sub-layers, and most especially their index, in particular by the choice of pyrolysis temperature of the precursors of this sub-layer.

20 Finally, the use of layers according to the invention is not impaired by the fouling known in previous methods. Therefore, we have particularly advantageous means, from the industrial point of view, for the production of the glazing units in question.



## PATENT CLAIMS:

1. Glazing comprising a glass substrate coated with a layer of aluminium oxynitride deposited by gas-phase pyrolysis, the thickness and refractive index characteristics thereof being selected so as to attenuate the reflected colours produced by an oxide layer providing the glazing with low-emission and/or solar protection properties, said layer being deposited onto the aluminium oxynitride layer.

2. Glazing according to Claim 1, wherein the constituent elements of the aluminium oxynitride layer are respectively in the following atomic proportions:

Al	from 40 to 50%
N	from 20 to 50%
O	from 10 to 60%.

3. Glazing according to Claim 2, wherein the constituent elements of the aluminium oxynitride layer are respectively in the following atomic proportions:

Al	from 45 to 50%
N	from 22 to 30%
O	from 20 to 27%.

4. Glazing according to one of the preceding claims, wherein the refractive index of the aluminium oxynitride layer is in the range of between 1.6 and 1.8.

5. Glazing according to Claim 4, wherein the refractive index of the aluminium oxynitride layer is in the range of between 1.65 and 1.75.

5 6. Glazing according to one of the preceding claims, wherein the thickness of the aluminium oxynitride layer has a thickness in the range of between 500 and 900 ångströms.

10 7. Glazing according to Claim 6, wherein the thickness of the aluminium oxynitride layer has a thickness in the range of between 650 and 850 ångströms.

15 8. Glazing according to one of the preceding claims, wherein the oxide layer providing the low-emission and/or solar protection properties is a layer based on doped tin oxide.

9. Glazing according to one of the preceding claims, wherein the oxide layer is a tin oxide layer containing antimony oxide, the atomic ratio Sb/Sn being in the range of between 0.02 and 0.15.

20 10. Glazing according to one of the preceding claims, wherein the oxide layer is a layer based on fluorine-doped tin oxide.

25 11. Process for the production of glazing according to one of the preceding claims, wherein the aluminium oxynitride layer is formed by pyrolysis using gaseous precursors comprising aluminium trichloride or trimethyl aluminium.

12. Process for the production of glazing according to Claim 10, wherein the gaseous precursors also comprise ammonia.

5 13. Process for the production of glazing according to one of Claims 11 or 12, wherein, when the aluminium precursor is aluminium chloride, the precursors also contain water vapour.

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Docket No.: 4004-028-30

**Declaration, Power of Attorney and Petition**

WE (I) the undersigned inventor(s), hereby declare(s) that:

My residence, post office address and citizenship are as stated below next to my name,

We (I) believe that we are (I am) the original, first, and joint (sole) inventor(s) of the subject matter which is claimed and for which a patent is sought on the invention entitled

**PYROLYTIC LAYER OF ALUMINIUM OXYNITRIDE AND GLAZING COMPRISING SAME**

the specification of which

- ☐ is attached hereto.
- ☒ was amended on 17 JANUARY 2002
- ☒ was filed as PCT international application

Number PCT/EP00/05615

on 19 JUNE 2000

and was amended under PCT Article 19

on \_\_\_\_\_ (if applicable).

We (I) hereby state that we (I) have reviewed and understand the contents of the above-identified specification, including the claims, as amended by any amendment referred to above.

We (I) acknowledge the duty to disclose information known to be material to the patentability of this application as defined in Section 1.56 of Title 37 Code of Federal Regulations.

We (I) hereby claim foreign priority benefits under 35 U.S.C. §119(a)-(d) or §365(b) of any foreign application(s) for patent or inventor's certificate, or §365(a) of any PCT International application which designated at least one country other than the United States, listed below and have also identified below, by checking the box, any foreign application for patent or inventor's certificate, or PCT International application having a filing date before that of the application on which priority is claimed. Prior Foreign Application(s)

Application No.	Country	Day/Month/Year	Priority Claimed
90420	LUXEMBOURG	20 JULY 1999	<input checked="" type="checkbox"/> Yes <input type="checkbox"/> No
_____	_____	_____	<input type="checkbox"/> Yes <input type="checkbox"/> No
_____	_____	_____	<input type="checkbox"/> Yes <input type="checkbox"/> No
_____	_____	_____	<input type="checkbox"/> Yes <input type="checkbox"/> No

10031237-020102

We (I) hereby claim the benefit under Title 35, United States Code, §119(e) of any United States provisional application(s) listed below.

(Application Number)

(Filing Date)

(Application Number)

(Filing Date)

We (I) hereby claim the benefit under 35 U.S.C. §120 of any United States application(s), or §365(c) of any PCT International application designating the United States, listed below and, insofar as the subject matter of each of the claims of this application is not disclosed in the prior United States or PCT International application in the manner provided by the first paragraph of 35 U.S.C. §112, I acknowledge the duty to disclose information which is material to patentability as defined in 37 CFR §1.56 which became available between the filing date of the prior application and the national or PCT International filing date of this application.

Application Serial No.

Filing Date

Status (pending, patented, abandoned)

And we (I) hereby appoint Steven B. Kelber, Reg. No. 30,073; Marc R. Labgold, Ph.D., Reg. No. 34,651; Jerold I. Schneider, Reg. No. 24,765; Paul C. Kimball, Reg. No. 34,641; Laura A. Donnelly, Reg. No. 38,435; Wilburn L. Chesser, Reg. No. 41,668; James M. Heintz, Reg. No. 41,828; Laura D. Nammo, Reg. No. 42,024; Perry E. VanOver, Reg. No. 42,197; Amy L. Miller, Reg. No. 43,804; Raymond Millien, Reg. No. 43,806; Lisa K. Norton, Reg. No. 44,977; Kenneth Vu, Reg. No. 46,323 and Christopher W. Raimund, Reg. No. 47,258, as our (my) attorneys, with full powers of substitution and revocation, to prosecute this application and to transact all business in the Patent Office connected therewith; and we (I) hereby request that all correspondence regarding this application be sent to Supervisor, Patent Prosecution Services, Piper Marbury Rudnick & Wolfe LLP, 1200 Nineteenth Street, N.W., Washington, D.C. 20036-2412.

We (I) declare that all statements made herein of our (my) own knowledge are true and that all statements made on information and belief are believed to be true; and further that these statements were made with the knowledge that willful false statements and the like so made are punishable by fine or imprisonment, or both, under Section 1001 of Title 18 of the United States Code and that such willful false statements may jeopardize the validity of the application or any patent issuing thereon.

Philippe LEGRAND

NAME OF SOLE INVENTOR

Residence: Chemin du Champ de Mars, 65

B-7000 Mons

BELGIUM

BEX

Citizen of: BELGIUM

Post Office Address: Same as above

Signature of Inventor

Date